SAFETY DATA SHEET (SDS)

1, 3-BUTADIENE

1. Identification

<table>
<thead>
<tr>
<th>SDS record number</th>
<th>PCS95003</th>
</tr>
</thead>
<tbody>
<tr>
<td>Date of SDS</td>
<td>01 October 2013</td>
</tr>
<tr>
<td>Identity of the substance</td>
<td>1, 3-Butadiene</td>
</tr>
<tr>
<td>Product Description</td>
<td>Di-olefin Hydrocarbon</td>
</tr>
<tr>
<td>Other names/synonyms</td>
<td>Biethylene, Divinyl, Butadiene, Divinyl, Erythrene, Vinylethylene</td>
</tr>
<tr>
<td>Name of the supplier</td>
<td>Petrochemical Corporation of Singapore (Private) Limited</td>
</tr>
<tr>
<td>Recommended uses</td>
<td>Chemical feedstock</td>
</tr>
<tr>
<td>Contact detail of the supplier</td>
<td>100 Ayer Merbau Road, Singapore 628277 +65 68672102</td>
</tr>
</tbody>
</table>

24-Hour Emergency contact:
- Asia Pacific: +65 3158 1074 (Singapore)
- China: +86 10 5100 3039 (Beijing)
- Europe, Israel & Americas: +44 (0) 1235 239 670 (UK)
- Middle East & Africa: +44 (0) 1235 239 671 (UK)

2. Hazards Identification

GHS Classification

<table>
<thead>
<tr>
<th>Hazard Class</th>
<th>Hazard Category</th>
</tr>
</thead>
<tbody>
<tr>
<td>Flammable Gas</td>
<td>1</td>
</tr>
<tr>
<td>Gases under pressure</td>
<td>Liquefied gas</td>
</tr>
<tr>
<td>Germ Cell Mutagenicity</td>
<td>1B</td>
</tr>
<tr>
<td>Carcinogenicity</td>
<td>1A</td>
</tr>
<tr>
<td>STOST (Single Exposure)</td>
<td>3 (respiratory tract irritation, narcotic effects)</td>
</tr>
<tr>
<td>STOST (Repeated Exposure)</td>
<td>1 (ovaries)</td>
</tr>
<tr>
<td></td>
<td>2 (blood system, heart, liver, bone marrow, testes)</td>
</tr>
</tbody>
</table>

Pictograms

Signal Word: Danger

Hazard Statements

- Extremely flammable gas
- Contains gas under pressure; may explode if heated
- May cause genetic defects
- May cause cancer
- May cause respiratory irritation
- May cause drowsiness and dizziness
- Causes damage to organs through prolonged or repeated exposure
Precautionary Statements

Prevention
- Keep away from heat/sparks/open flames/hot surfaces. – No smoking.
- Obtain special instructions before use.
- Do not handle until all safety precautions have been read and understood.
- Use personal protective equipment as required.
- Use only outdoors or in well-ventilated area.
- Avoid breathing dust/fume/gas/mist/vapours/spray.
- Do not eat, drink or smoke when using this product.
- Wash hands thoroughly after handling.

Response
- Leaking gas fire: Do not extinguish, unless leak can be stopped safely.
- Eliminate all ignition sources if safe to do so.
- IF exposed or concerned: Get medical attention/advice.
- Call a POISON CENTER/doctor/physician if you feel unwell.
- If INHALED: If breathing is difficult, remove victim to fresh air and keep at rest in a position comfortable for breathing.

Storage
- Store in well-ventilated place. Keep container tightly closed.
- Store locked up.
- Protect from sunlight. Store in well-ventilated place.

Disposal
- Dispose of the contents in accordance to the local mandatory rules and regulations

3. Composition/Information On Ingredients
Chemical identification: \( \text{CH}_2=\text{CHCH}=\text{CH}_2 \)
Common name(s) / synonym(s): Biethylene, Divinyl, Butadiene, Divinyl, Erythrene, Vinylethylene
CAS number / EC number: 106-99-0/203-450-8

4. First-Aid Measures

Ingestion: No emergency care anticipated. This material is a gas at standard temperature and pressure.

Eye: If frostbite has occurred, seek medical attention immediately; if tissue is not frozen, immediately and thoroughly flush the eyes with large amounts of water for at least 15 minutes, occasionally lifting the lower and upper eyelids. If irritation, pain, swelling, lacrimation, or photophobia persists, get medical attention as soon as possible.

Skin: If frostbite has occurred, seek medical attention immediately; do not rub the affected areas or flush them with water. In order to prevent further tissue damage, do not attempt to remove frozen clothing from frostbitten areas. If frostbite has not occurred, immediately and thoroughly wash contaminated skin with soap and water. In case of cold burn, immediately place affected area in warm water (105 F) and keep at this temperature until circulation returns. Get medical attention.

Breathing: Respiratory support. If a person breathes large amounts of this chemical, move the exposed person to fresh air at once. If breathing has stopped, perform artificial resuscitation. Keep the affected person warm and at rest. Get medical attention as soon as possible.
Other Instructions: Clothing contaminated with liquefied flammable gases may give rise to delayed evaporation and cause a subsequent fire hazard. Drench individual with water and remove contaminated clothing if possible. Slowly warm affected area of skin. Do not attempt to reheat rapidly.

5. Fire-Fighting Measures

Recommended Fire Extinguishing Agents And Special Procedures:
- Fight fire from protected location or maximum possible distance.
- Stop flow of gas before attempting to extinguish flames.
- Use water spray to cool fire-exposed containers and to protect persons attempting to stop the flow of gas.
- Use flooding quantities of water as fog or spray.
- Use dry chemical or carbon dioxide to extinguish flames.

Unusual or Explosive Hazards:
- Danger Extremely flammable.
- Flashback may occur along vapour trail.
- May form explosive peroxides on exposure to air.
- Containers may explode in fire.

Special Protective Equipment for Fire-fighters: Wear full protective clothing and positive pressure breathing apparatus.

6. Accidental Release Measures

Evacuate danger area! Consult an expert!

Never direct water jet on liquid. Remove all ignition sources. Chemical protection suit including self-contained breathing apparatus.

Procedures in Case of Accidental Release, Breakage or Leakage:
- Eliminate all sources of ignition.
- Ventilate area of spill or leak.
- Stop flow of liquid at source if possible.
- Prevent liquid from entering sewers.
- Dilute with water fog.
- Keep people away.
- Stay upwind and warn of possible downwind explosion hazard.
- Avoid breathing vapour. Avoid contact with eyes, skin, and clothing.
- Pressure demand air supplied respirators should always be worn when the airborne concentration of the contaminant or oxygen is unknown. Otherwise, wear respiratory protection and other personal protective equipment as appropriate for the potential exposure hazard.

7. Handling And Storage

Precautions For Safe Handling
- Use spark-proof tools.
- Do not contact with copper or copper alloys as explosive copper compounds may be formed.
- Piping material for this gas must not contain over 63% of copper.

Storage:
- Ground and bond shipping container, transfer line, and receiving container.
- Keep away from heat, sparks, flame, and other sources of ignition.
- Protect containers against static electricity, lightning, and physical damage.
- Must be kept inhibited during storage and shipment.
- Closed system, ventilation, explosion-proof electrical equipment and lighting.
- Prevent build-up of electrostatic charges (e.g., by grounding) if in liquid state.
8. Exposure Controls/Personal Protection

Appropriate Engineering Controls
Ventilation: Use explosion-proof equipment to maintain adequate ventilation to meet occupational exposure limits, if applicable (see below), prevent accumulation of explosive air-gas mixtures, and avoid significant oxygen displacement.
Oxygen levels should be at least 19.5% in confined spaces or other work areas (OSHA value).

Personal Protective Equipment (PPE)

Eye/Face Protection: Safety glasses, chemical type goggles, or face shield recommended to prevent eye contact. Wear appropriate eye protection to prevent eye contact with the liquid that could result in burns or tissue damage from frostbite.

Skin Protection: Protective clothing such as coveralls or lab coats should be worn. Launder or dry-clean when soiled. Gloves and boots resistant to chemicals and petroleum distillates required. Insulated gloves also required if contact with liquid-cooled product or equipment is expected. Wear appropriate personal protective clothing to prevent the skin from becoming frozen. Leaks and uses that allow rapid expansion may cause a frostbite hazard.

Respiratory Protection: Airborne concentrations should be kept to lowest levels possible. If vapor, mist or dust is generated and the occupational exposure limit of the product, or any component of the product, is exceeded, use appropriate.

Remove: When wet (flammable) Work clothing that becomes wet should be immediately removed due to its flammability hazard (i.e., for liquids with a flash point <100°F).

Provide: Frostbite wash Quick drench facilities and/or eyewash fountains should be provided within the immediate work area for emergency use where there is any possibility of exposure to liquids that are extremely cold or rapidly evaporating.

Respirator Recommendations from NIOSH

<table>
<thead>
<tr>
<th>Airborne Concentration or Condition of Use</th>
<th>Required Respirator</th>
</tr>
</thead>
<tbody>
<tr>
<td>&lt; or = 5 ppm (parts per million)</td>
<td>Air-purifying half-mask or full-face piece respirator equipped with approved butadiene or organic vapor cartridges or canisters. Cartridges or canisters shall be replaced every 4 hours.</td>
</tr>
<tr>
<td>&lt; or = 10 ppm</td>
<td>Air-purifying half-mask or full-face piece respirator equipped with approved butadiene or organic vapor cartridges or canisters. Cartridges or canisters shall be replaced every 3 hours.</td>
</tr>
<tr>
<td>&lt; or = 25 ppm</td>
<td>(1) Air-purifying half-mask or full-face piece respirator equipped with approved butadiene or organic vapor cartridges or canisters. Cartridges or canisters shall be replaced every 2 hours; (2) Any powered air-purifying respirator equipped with approved butadiene or organic vapor cartridges or canisters. Cartridges or canisters shall be replaced every [1] hour; or (3) Continuous-flow supplied-air respirator equipped with a hood or helmet.</td>
</tr>
<tr>
<td>&lt; or = 50 ppm</td>
<td>(1) Air-purifying full-face piece respirator equipped with approved butadiene or organic vapor cartridges or canisters. Cartridges or canisters shall be replaced every [1] hour; or (2) Powered air-purifying respirator (PAPR) equipped with a tight-fitting face piece and approved butadiene or organic vapor cartridges. PAPR cartridges shall be replaced every [1] hour.</td>
</tr>
</tbody>
</table>
1, 3-Butadiene SDS
Page 5 of 8

<table>
<thead>
<tr>
<th>Concentration Level</th>
<th>Recommended Respiratory Protection</th>
</tr>
</thead>
<tbody>
<tr>
<td>&lt; or = 1,000 ppm</td>
<td>Supplied-air respirator equipped with a half-mask or full face piece and operated in a pressure-demand or other positive-pressure mode.</td>
</tr>
<tr>
<td>&gt; 1,000 ppm, unknown concentration, or firefighting</td>
<td>(1) Self-contained breathing apparatus equipped with a full face piece and operated in a pressure-demand or other positive-pressure mode; or (2) Any supplied-air respirator equipped with a full face piece and operated in a pressure-demand or other positive-pressure mode in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode.</td>
</tr>
<tr>
<td>Escape from IDLH conditions (IDLH is 2,000 ppm)</td>
<td>(1) Any positive-pressure self-contained breathing apparatus with an appropriate service life; or (2) Any air-purifying full-face piece respirator equipped with a front- or back-mounted butadiene or organic vapor canister.</td>
</tr>
</tbody>
</table>

### 9. Physical And Chemical Properties

<table>
<thead>
<tr>
<th>Property</th>
<th>Value, Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>Appearance (physical state, colour etc);</td>
<td>Compressed gas Colourless gas</td>
</tr>
<tr>
<td>Odour;</td>
<td>Aromatic odour</td>
</tr>
<tr>
<td>Odour threshold;</td>
<td>Not available</td>
</tr>
<tr>
<td>pH;</td>
<td>Not applicable.</td>
</tr>
<tr>
<td>Melting point;</td>
<td>-109°F</td>
</tr>
<tr>
<td>Initial boiling point and boiling range;</td>
<td>-4 deg C 23.5 F</td>
</tr>
<tr>
<td>Flash point;</td>
<td>-76°F</td>
</tr>
<tr>
<td>Evaporation rate;</td>
<td>Not available</td>
</tr>
<tr>
<td>Upper/lower flammability or explosive limits;</td>
<td>2% to 12% Vol</td>
</tr>
<tr>
<td>Vapour pressure;</td>
<td>1810 mmHg at 68.0</td>
</tr>
<tr>
<td>Vapour density;</td>
<td>2.36 kg/m3 @ 1.013 bar 15 deg C</td>
</tr>
<tr>
<td>Relative density;</td>
<td>1.92 @1.013 bar 21 deg C (Air = 1)</td>
</tr>
<tr>
<td>Solubility(ies);</td>
<td>2 g/l (water, 20 deg C) soluble (organic solvents, 20 deg C)</td>
</tr>
<tr>
<td>Partition coefficient: n-octanol/water;</td>
<td>1.99</td>
</tr>
<tr>
<td>Auto-ignition temperature;</td>
<td>804°F (415°C)</td>
</tr>
<tr>
<td>Decomposition temperature;</td>
<td>Not available</td>
</tr>
<tr>
<td>Viscosity.</td>
<td>Not applicable.</td>
</tr>
</tbody>
</table>

### 10. Stability And Reactivity

**Reactivity/Chemical stability:** This material reacts with Heat, Strong Oxidizers, and reducing agents, copper, and copper alloys.

**Possibility of hazardous reactions:** May spontaneously dimerize to 4-vinyl-1-cyclohexene in a temperature-dependent reaction. Vapors may form polymers which block vents or flame arrestors.

**Incompatible materials:** Phenol, chlorine dioxide, copper, crotonaldehyde.

**Hazardous decomposition products:** Carbon monoxide and carbon dioxide may be formed on burning in a limited air supply.

**Hazardous Polymerizations:** May Occur

**Conditions to avoid:** Avoid proximity or contact with hot surfaces, flames, electrostatic charges or sparks. Do not store at a temperature exceeding 25 °C.

**Note:** May form explosive peroxides upon exposure to air.
11. Toxicological Information

Product is a gas - not expected to be absorbed through the skin. Skin contact with liquid product can cause frostbite (cold burns).

**Inhalation:** Gas may be irritating and cause discomfort in nose and throat, nasal discharge, and coughing. Prolonged overexposure may cause difficulty breathing. Inhalation may cause dizziness, drowsiness, euphoria, loss of coordination, disorientation, headache, nausea, and vomiting. In poorly ventilated areas or confined spaces, unconsciousness and asphyxiation may result. Prolonged or repeated overexposure may result in the absorption of potentially harmful amounts of material.

**Ingestion:** Product is a gas - not expected to cause toxic effects due to ingestion.

This material is a gas. Gas or liquid under pressure may cause frostbite (cold burns).

**Sensitization Properties:** Unknown.

**Toxicological Information (Animal Toxicity Data)**

**Median Lethal Dose**

**Oral:** LD50 Not applicable; material is a gas.

**Inhalation:** 4 hr. LC50 > 99000.00 ppm (gas, vapor) (rat) practically non-toxic

**Dermal:** LD50 Not applicable; material is a gas.

**Irritation Index:** Estimation of Irritation (Species)

**Skin:** (Draize) Believed to be < .50 /8.0 (rabbit) no appreciable effect

**Eyes:** (Draize) Believed to be > 15.00 - 25.00 /110 (rabbit) slightly irritating

**Sensitization:** Not determined.

The current IARC classification of 1,3-butadiene is Class 2A, meaning that IARC finds there is "limited evidence" of 1,3-butadiene carcinogenicity in humans and "sufficient evidence" of 1,3-butadiene carcinogenicity in experimental animals.

**Medical Conditions Aggravated by Exposure:** There is no evidence that this product aggravates an existing medical condition.

**Other:** Prolonged and repeated inhalation of 1, 3-butadiene has produced tumours in multiple sites in rats and mice. In Sprague-Dawley rats exposed to 1000 or 8000 ppm butadiene, tumour sites have included the mammary gland, thyroid, and testes. The National Toxicology Program concluded there was "clear evidence" of carcinogenicity in B6C3F1 mice exposed to 6.25 ppm to 1250 ppm butadiene. This was based on increased tumours in the haematopoietic system, heart, lung, fore stomach, liver and harderian gland in males and females, prenuptial gland, brain, and kidney (males), and in ovary and mammary gland (females). IARC has concluded that there is sufficient evidence for 1, 3-butadiene carcinogenicity in experimental animals.

Repeated exposure to 1, 3-butadiene has produced genetic toxicity, bone marrow toxicity, and anaemia in the mouse. No carcinogenic damage to the ovary, testes, liver, nasal tissue, and fore stomach have also been observed in the mouse, and evidence of kidney damage has been observed in the rat. Exposure of pregnant rodents to maternally toxic 1, 3-butadiene concentrations has affected the developing foetus. Malformations (birth defects) have been reported in the developing foetus of pregnant rats exposed to 8000 ppm 1, 3-butadiene. There was no evidence of teratogenic effects in a second developmental study in the rat or a developmental study in the mouse, both involving butadiene exposures up to 1000 ppm.
The B6C3F1 mouse has been demonstrated to be substantially more susceptible to toxic and carcinogenic responses to 1, 3-butadiene exposure than the Sprague-Dawley rat. Repeated exposure to 6.25 ppm 1, 3-butadiene has produced lung tumours and ovarian atrophy in females, and evidence of genetic toxicity in males and females of this mouse strain. By contrast, evidence for toxic and carcinogenic responses in the rat is more limited and has been observed primarily following prolonged exposure to 1, 3-butadiene concentrations of 1000 ppm or higher.

In an effort to explain the higher toxic and carcinogenic potency of 1,3-butadiene in the mouse and evaluate the relevance of these animal bioassay results to humans, pharmacokinetic and metabolism studies have been conducted using rodents, monkeys, and tissues from rodents and humans. The results of these studies suggest that the mouse may not be an appropriate model from which to predict human health effects from exposure to 1, 3-butadiene.

Chronic: A study conducted at the Texaco Chemical Company 1,3-butadiene manufacturing facility (subsequently acquired by Huntsman Corporation) and updated through 1990 showed that workers had a lower overall death rate than the general U.S. population. There was a two-fold increase in mortality from lymph sarcoma and reticulum cell sarcoma. However, the increase was not consistently seen in all job groups potentially exposed to higher levels of 1, 3-butadiene. In addition, the increase was seen primarily in short term employees whose employment began during World War II and not in those employed ten years or more. This pattern of results suggests that exposure to 1; 3-butadiene was not responsible for the increased lymph sarcomas. Several other studies involving workers manufacturing styrene-butadiene rubber have shown increases of leukaemia and lymphomas. These workers were potentially exposed to 1, 3-butadiene and other chemicals, especially styrene and possibly benzene. Due to the presence of multiple chemicals and inconsistent results, these studies also do not demonstrate 1, 3-butadiene exposure to be responsible for increased leukaemia or lymphoma in these workers.

12. Ecological Information

Acute Toxicity

**Fish:** Expected to be harmful: \(10 < \text{LC/EC/IC50} \leq 100 \text{ mg/l}\)

**Aquatic Invertebrates:** Expected to be harmful: \(10 < \text{LC/EC/IC50} \leq 100 \text{ mg/l}\)

**Algae:** Expected to be harmful: \(10 < \text{LC/EC/IC50} \leq 100 \text{ mg/l}\)

**Mobility:** Floats on water.

**Persistence/degradability:** Expected to be inherently biodegradable. Oxidizes rapidly by photochemical reactions in air.

**Bioaccumulation:** Does not bioaccumulate significantly.

13. Disposal Considerations

Recover or recycle if possible. It is the responsibility of the waste generator to determine the toxicity and physical properties of the material generated to determine the proper waste classification and disposal methods in compliance with applicable regulations.

Disposal should be in accordance with applicable regional, national, and local laws and regulations.

14. Transport Information

UN Number: UN 1010

UN Proper Shipping Name: Butadiene

Hazard Class: 2.1

Label Required: Flammable gas
Do not transport with food and feedstuffs.
Note: D
F+ symbol
T symbol
R: 45-46-12
S: 53-45
UN Hazard Class: 2.1
Transport Emergency Card: TEC (R)-20S1010
NFPA Code: H2; F4; R2;

IMDG
Identification number UN 1010
Proper shipping name BUTADIENES, STABILIZED
Class / Division 2.1
Marine pollutant: No

IATA (Country variations may apply)
UN No.: 1010
Proper shipping name: Butadienes, stabilized
Class / Division: 2.1

15. Regulatory Information

Chemical Name 1, 3-butadiene
CAS Number 106-99-0 95
Concentration.00-99.99
Permissible Exposure Level (Long Term) in Singapore: 2ppm (4.4mg/m³)

WHMIS Classification:
Class A: Compressed gas
Class B, Div 1: Flammable gas
Class D, Div 2, Subdiv A: Carcinogenic

TLV: 2 ppm as TWA; A2 (suspected human carcinogen); (ACGIH 2004).
Carcinogen category: 1; Germ cell mutagen group: 2;
(DFG 2004).
OSHA PEL: 1910.1051 TWA 1 ppm ST 5 ppm
NIOSH IDLH: Potential occupational carcinogen 2000 ppm LEL

16. Other Information

Prepared By: Material Safety Committee
SDS Prepared on: 1/10/2010

CAUTION: The information given above ("the Information") relates only to the substance or mixture listed herein. The Information may not be valid when used in combination with any other substance or mixture or in any process. If the substance or mixture is to be used for a purpose other than that stated herein or under conditions other than specified herein, the Information cannot be relied upon as being complete or accurate, and the user is advised to consult the supplier before using the substance or mixture for such other purpose or under such other conditions. The Information is given based on information available at the indicated date of preparation and no representation or warranty is given that it will be correct as of any time after the indicated date of preparation.